

Provided for non-commercial research and education use.  
Not for reproduction, distribution or commercial use.



This article appeared in a journal published by Elsevier. The attached copy is furnished to the author for internal non-commercial research and education use, including for instruction at the authors institution and sharing with colleagues.

Other uses, including reproduction and distribution, or selling or licensing copies, or posting to personal, institutional or third party websites are prohibited.

In most cases authors are permitted to post their version of the article (e.g. in Word or Tex form) to their personal website or institutional repository. Authors requiring further information regarding Elsevier's archiving and manuscript policies are encouraged to visit:

<http://www.elsevier.com/copyright>



Contents lists available at ScienceDirect

## Tetrahedron Letters

journal homepage: [www.elsevier.com/locate/tetlet](http://www.elsevier.com/locate/tetlet)

## Synthesis of deuterated 5(Z),11(Z)-eicosadienoic acid as a biomarker for trophic transfer

Siliva Albu<sup>a</sup>, Ed Sverko<sup>b</sup>, Michael T. Arts<sup>b</sup>, Alfredo Capretta<sup>a,\*</sup>

<sup>a</sup> Department of Chemistry and Chemical Biology, McMaster University, Hamilton, Ontario, Canada L8S 4M1

<sup>b</sup> Water Science and Technology Directorate, National Water Research Institute, Environment Canada, 867 Lakeshore Road, PO Box 5050, Burlington, Ontario, Canada L7R 4A6

## ARTICLE INFO

## Article history:

Received 28 October 2010

Revised 2 December 2010

Accepted 6 December 2010

Available online 17 December 2010

## ABSTRACT

The poly-methylene interrupted fatty acid 5(Z),11(Z)-eicosadienoic acid and its tetradeuterated analogue were prepared via a convergent synthetic sequence.

© 2010 Elsevier Ltd. All rights reserved.

Fatty acids (FAs) are recognized as important tools in ecology,<sup>1</sup> toxicology<sup>2</sup> and conservation biology.<sup>3</sup> While they have been utilized previously as trophic markers,<sup>4–6</sup> their use is hampered by the fact that they exhibit species-specific bioconversion rates and also because many FAs are ubiquitous and abundant in aquatic food webs. In contrast, poly-methylene interrupted fatty acids<sup>7</sup> (PMI-FAs) possess an unusual methylene substitution pattern and, as such, occur much less frequently in nature. In the marine environment, it is thought that PMI-FAs are synthesized de novo primarily in bivalves and carnivorous gastropods<sup>8–10</sup> and further accumulate, to varying extents (depending on diet), in marine mammals.<sup>9</sup>

Our interest in PMI-FAs stems from their potential as biomarkers for zebra (*Dreissena polymorpha*) and quagga (*D. bugensis*) mussels (termed dreissenids). Since their introduction in the Laurentian Great Lakes, their adaptability, rapid life cycle, and high reproductive potential have ensured their continued invasion success<sup>11,12</sup> and has resulted in extensive ecosystem impact and damage.<sup>13,14</sup> New methods designed to enable researchers to better track the passage of dreissenid-derived carbon to both native and introduced consumers at different trophic levels through the food web are required in an effort to understand the extent to which dreissenids alter the function and stability of aquatic ecosystems. The PMI-FA 5(Z),11(Z)-eicosadienoic acid (**1**) was identified as a useful biomarker for these studies and the present paper describes the synthesis of a deuterated analogue applicable for analytical studies using isotopic dilution.<sup>15</sup>

Despite having been isolated from a variety of natural sources, the synthesis of 5(Z),11(Z)-eicosadienoic acid (**1**) has not been described in the literature. The synthetic route developed appears in Schemes 1 and 2. The approach was designed so as to insure *cis* stereochemistry at C5=C6 and C11=C12 by taking advantage of an al-

kyne reduction using Lindlar's catalyst. Furthermore, the route offers the opportunity to introduce deuterium at either alkene or both as desired.

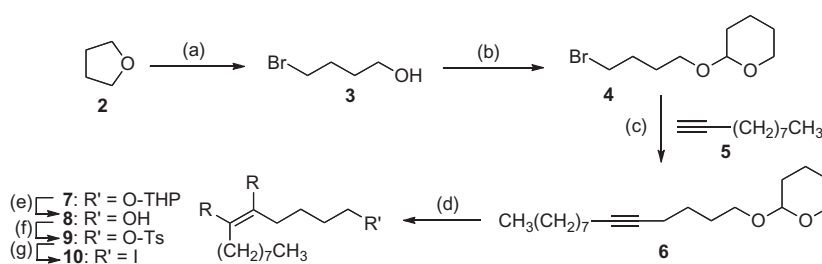
Initial development of the synthesis focused on preparation of the non-deuterated fatty acid. The route to the C7–C20 fragment appears in Scheme 1 (R = H). While 4-bromo-1-butanol (**3**) is commercially available, a convenient synthesis has been described previously wherein tetrahydrofuran (**2**) is treated with HBr.<sup>16</sup> The resulting bromo-alcohol was then protected as its THP ether (**4**)<sup>17</sup> before coupling with 1-decyne (**5**) in THF at –78 °C to give **6**. Heterogeneous hydrogenation of the alkyne to the alkene (**7**) is carried out in the presence of Lindlar's catalyst. Inspection of the <sup>1</sup>H NMR (at 700 MHz) allowed for resolution of the alkenic protons and revealed only a single isomer. Furthermore, coupling of the alkenic protons to each other with a *J* = 6.9 Hz was consistent with the desired *cis* geometry. Removal of the THP protecting group affords the free alcohol **8** that was subsequently tosylated (to give **9**). Finally, treatment with NaI in acetone provides **10**, the required C7–C20 synthon.

Synthesis of the C1–C6 synthon along with its coupling to the C7–C20 fragment appears in Scheme 2 (R = H). Commercially available 5-hexyn-1-ol (**11**) was protected as its THP ether (**12**)<sup>18</sup> before coupling to iodo-alkene **10** to yield alkene-alkyne **13**. It should be noted that attempts to couple **12** with the tosylate **9** failed prompting the use of the iodo compound **10**. Reduction to the bisalkene (**14**) once again took advantage of the H<sub>2</sub>/Lindlar's catalyst system with *cis* geometry at the newly formed C5=C6 alkene confirmed via <sup>1</sup>H NMR. Deprotection of the THP ether **14** to the alcohol **15** is followed by oxidation using pyridinium dichromate in DMF to yield the desired 5(Z),11(Z)-eicosadienoic acid (**1**; R = H).

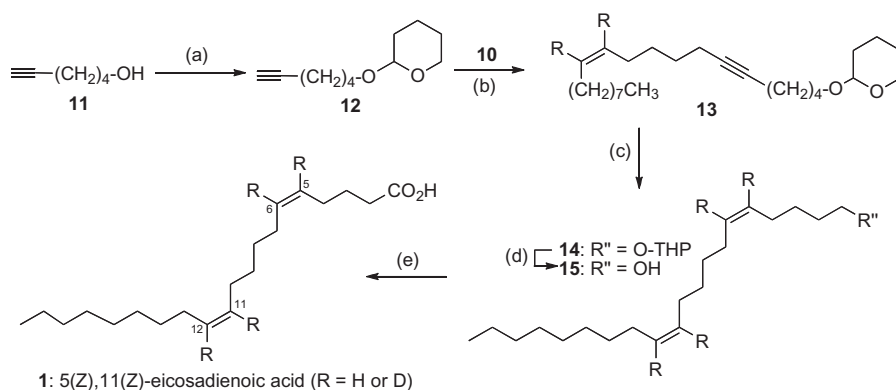
Having established the viability of the synthetic route, attention was turned to the preparation of the deuterated analogue (Schemes 1 and 2 where R = D). Use of D<sub>2</sub> gas in place of H<sub>2</sub> in the hydrogenations of **6** to **7** and **13** to **14** allowed for the incorporation of deuterium into the newly generated *cis* alkenes. [<sup>2</sup>H<sub>4</sub>]-

\* Corresponding author. Tel.: +1 905 525 9140x27318; fax: +1 905 522 2509.

E-mail address: [capretta@mcmaster.ca](mailto:capretta@mcmaster.ca) (A. Capretta).



**Scheme 1.** Synthesis of the C7–C20 synthon. Reagents and conditions: (a) HBr, H<sub>2</sub>SO<sub>4</sub> (90%); (b) DHP, *p*-TsOH, DCM (72%); (c) (i) **5**, *n*-BuLi, THF, –78 °C; (ii) dropwise addition of **4** then heat (85%); (d) 1 atm H<sub>2</sub>, Lindlar's catalyst (72%); (e) *p*-TsOH, MeOH (68%); (f) *p*-TsCl, DCM, pyridine (73%); (g) NaI, acetone (71%).



**Scheme 2.** Synthesis of the C1–C6 synthon and coupling to the C7–C20 fragment. Reagents and conditions: (a) DHP, *p*-TsOH, DCM (92%); (b) (i) **12**, *n*-BuLi, HMPA, THF, –78 °C; (ii) dropwise addition of **10** then heat (65%); (c) 1 atm H<sub>2</sub>, Lindlar's catalyst (62%); (d) *p*-TsOH, MeOH (92%); (e) PDC, DMF (45%).

5(*Z*), 11(*Z*)-Eicosadienoic acid (**1**; R = D) was completely characterized by <sup>1</sup>H NMR, <sup>13</sup>C NMR and HRMS.<sup>19</sup> Furthermore, gas chromatographic analysis of the deuterated PMI-FA with an authentic standard revealed that both samples had similar retention times.<sup>20</sup>

In conclusion, 5(*Z*),11(*Z*)-eicosadienoic acid and its tetra-deuterated analogue were prepared via a convergent 12-step synthesis. The nature of the route is such that other PMI-FAs can be prepared in this fashion. Work describing the use of the deuterated standard as a biomarker for trophic transfer will be described elsewhere.

## Acknowledgements

The authors thank the Natural Sciences and Engineering Research Council of Canada, Environment Canada, the Canadian Foundation for Innovation and the Ontario Innovation Trust for their financial support.

## References and notes

- Lipids in Aquatic Ecosystems*; Arts, M. T., Brett, M. T., Kainz, M. J., Eds.; Springer: New York, 2009.
- Kainz, M. J.; Fisk, A. T. In *Lipids in Aquatic Ecosystems*; Arts, M. T., Brett, M. T., Kainz, M. J., Eds.; Springer: New York, 2009; pp 237–255.
- Glayshev, M. I.; Arts, M. T.; Sushchik, N. N. In *Lipids in Aquatic Ecosystems*; Arts, M. T., Brett, M. T., Kainz, M. J., Eds.; Springer: New York, 2009; pp 237–255.
- Koussoroplis, A.-M.; Lemarchand, C.; Bec, A.; Desvilettes, C.; Amblard, C.; Fournier, C.; Berny, P.; Bourdier, G. *Lipids* **2008**, *43*, 461–466.
- Napolitano, G. In *Lipids in Freshwater Ecosystems*; Arts, M. T., Wainman, B. C., Eds.; Springer: New York, 1999; pp 21–44.
- Desvilettes, C.; Bourdier, G.; Amblard, C.; Barth, B. *Freshwater Biol.* **1997**, *38*, 629–637.
- Mezek, T.; Arts, M. T.; Sverko, E.; Fisk, A. T. *Verh. Int. Ver. Limnol.* **2009**, *30*, 903–906.
- Saito, H. *J. Chromatogr., A* **2007**, *1163*, 247–259.
- Budge, S. M.; Springer, A. M.; Iverson, S. J.; Sheffield, G. *Mar. Ecol. Prog. Ser.* **2007**, *336*, 305–309.
- Budge, S. M.; Iverson, S. J.; Koopman, H. N. *Mar. Mamm. Sci.* **2006**, *22*, 759–801.
- Schloesser, D. W.; Metcalfe-Smith, J. L.; Kovalak, W. P.; Longton, G. D.; Smithee, R. D. *Am. Midl. Nat.* **2009**, *155*, 307–320.
- Zanatta, D. T.; Mackie, G. L.; Metcalfe-Smith, J. L.; Woolnough, D. A. *J. Great Lakes Res.* **2002**, *28*, 479–489.
- Connelly, N.; O'Neill, C.; Knuth, B.; Brown, T. *Environ. Manage.* **2007**, *40*, 105–112.
- Karatayev, A. Y.; Boltovskoy, D.; Padilla, D. K.; Burlakova, L. E. *J. Shellfish Res.* **2009**, *26*, 205–213.
- Hamon, R. E.; Parker, D. R.; Lombi, E.; Donald, L. S. In *Advances in Agronomy*; Academic Press, 2008; pp 289–343.
- Nguyen, J. T.; McEwen, C. A.; Knaus, E. E. *Drug Dev. Res.* **2000**, *51*, 233–243.
- Snider, B. B.; Lu, Q. *J. Org. Chem.* **1996**, *61*, 2839–2844.
- Gu, H.; Xu, W. M.; Kinstle, T. H. *Tetrahedron Lett.* **2005**, *46*, 6449–6451.
- <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>): 2.36 (t, 2H, *J* = 7 Hz), 2.09 (t, 2H, *J* = 7 Hz), 2.01–1.94 (m, 6H), 1.69 (m, 2H), 1.30–1.26 (m, 16H), 0.87 (t, 3H, *J* = 7 Hz); <sup>13</sup>C NMR (120 MHz, CDCl<sub>3</sub>): 178.8, 131.2, 131.0, 129.6, 128.2, 33.3, 32.0, 29.9, 2 × 29.6, 2 × 29.5, 29.4, 27.2, 27.1, 27.1, 26.4, 24.7, 22.8, 14.2; ES-HRMS: Calculated for C<sub>20</sub>H<sub>31</sub>O<sub>2</sub>D<sub>4</sub> [M–H]<sup>–</sup> 311.2888; observed: 311.2884.
- The PMI-FAs were analysed by GC–MS (Agilent 6890N GC) equipped with a DB-23 polar capillary column (Agilent; #122-2361; 60 m × 0.25 mm id × 0.15 μm film thickness), an Agilent 7683B injector, and a mass selective quadrupole detector (Agilent 5973N). Helium was used as the carrier gas at a constant pressure (~180 kPa at 33 cm s<sup>–1</sup> at 50 °C). Samples were injected at an oven temperature of 50 °C. After 1 min, the oven temperature was raised to 175 °C at a rate of 25 °C min<sup>–1</sup>, then to 235 °C at 4 °C min<sup>–1</sup> and held for 5 min. The retention times for 5(*Z*),11(*Z*)-eicosadienoic acid and [<sup>2</sup>H<sub>4</sub>]-5(*Z*),11(*Z*)-eicosadienoic acid were 18.1 and 18.2 min, respectively.